

TARGETING NITROSAMINE PRECURSORS WITH NOVEL ACTIVE MATERIAL SCIENCE SCAVENGER TECHNOLOGIES

In this piece, Ivy Comer, Scientist III, Amanda Murph, Scientist II, and Jason Pratt, Director, Material Science, all at Aptar CSP Technologies, discuss the challenges in combatting N-nitrosamine formation in oral solid dosage forms, and present the company's 3-Phase Activ-Polymer technology as a novel packaging-based solution.

N-nitrosamine or nitrosamine drug substance-related impurities (NDSRIs) are a class of potentially carcinogenic compounds that have been wreaking havoc on the pharmaceutical industry, leading to product recalls and heightened regulatory scrutiny of both commercialised and in-development drugs. Nitrosamines are found as impurities in various drug products, and studies show that, over time, nitrosamine formation levels can grow beyond the US FDA's daily thresholds, potentially putting patients at risk.

Nitrosamines are substances that can form under certain conditions through a process known as nitrosation, especially in the presence of nitrosating agents such as nitrite salts, alkyl nitrites, nitrogen oxides (NO_{x}), nitrosyl halides, nitrosonium salts and nitro compounds, along with amines or amides. Nitrosating agents can arise from the use of recycled solvents or reusing catalysts across different processes

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or manufacturing lines with inadequate control and inappropriate monitoring. They can also be generated from impure starting materials or intermediates in the upstream step or from the carry-over of other manufacturing processes along the same production line.

Another key variable in nitrosamine formation is the nitrite levels in many of the most commonly used excipients – for example, magnesium stearate and microcrystalline cellulose (MCC) can average 2,000–3,000 ppb levels of nitrites, aiding in nitrosamine-forming reactions.

Controlling the presence of nitrosating agents and their interaction with precursors is crucial in mitigating nitrosamine formation. Aptar CSP Technologies has conducted a series of studies1 that explores the possibility of leveraging innovative active material science technologies to develop a novel nitrosamine-mitigating solution for oral solid dosage forms. The branch of the study presented here focuses on the role that nitrites in excipients play in nitrosamine formation (Box 1). It also investigates how Aptar CSP Technologies' proprietary Activ-PolymerTM platform technology can mitigate the risk associated with MCC. The data presented explore whether nitrite impurity levels contained in a sample lot of placebo MCC tablets can be meaningfully reduced when placed together with Aptar CSP's Activ-FilmTM material in a sealed environment.

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BOX 1: THE IDEA IN BRIEF

The Problem

Pharma companies are facing increasing regulatory scrutiny and higher risk of product recalls due to N-nitrosamines and NDSRIs in drug products. The mechanism of reaction and root cause of nitrosamine formation is not always understood, which makes mitigating this risk difficult.

The Challenge

The current mitigation strategies employed by pharma developers, such as making changes to their formulation or manufacturing processes, can be costly. Additionally, it is difficult to control the variables associated with N-nitrosamine formation with any certainty. Additional or alternative mitigation strategies may need to be considered to fully address the risk.

The Solution

New innovations in active material science technologies can offer an active packaging-based solution that can react with NO_{x} gases in the packaging headspace to inhibit nitrosamine formation. Not only can these technologies stop nitrosamine formation, but they can serve as an additional "insurance policy" by adsorbing or scavenging N-nitrosamines post-formation, providing a holistic risk mitigation solution.

MITIGATING NITROSAMINE RISKS ASSOCIATED WITH EXCIPIENTS

Oral drug formulations typically involve a blend or granulation of their APIs with various excipients, many of which have been shown to include nitrite impurities at trace levels. According to Boetzel et al, the excipients, such as fillers and diluents, have the greatest influence over the nitrite concentration.2 MCC is an extremely common diluent in oral solid dosage forms, particularly tablets and capsules, and has the highest number of results in Lhasa Limited's (Leeds, UK) "Nitrite in Excipients" database at 0.04-2.4 ppm. These nitrites can form nitrosating species, such as nitrous acid and various volatile NO_v compounds, especially under moderately acidic conditions. For APIs at risk of nitrosamine formation, the presence of nitrosating species in the excipients used in the formulation has been shown to directly increase the rate of formation of nitrosamines.

As pharmaceutical developers seek to assess and mitigate the risk of N-nitrosamine formation in their current drug products and in-development APIs, they have employed strategies such as using nitrocellulose-free packaging or choosing alternate or supplemental excipients, reagents or catalysts. However, these solutions are not infallible, as companies may not always be able to source low- or no-nitrite excipients, and these strategies can result in overformulation, which could potentially lead to new problems.

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Additionally, the mechanism of reaction and root cause of nitrosamine formation is not always fully understood, complicating the process of finding an adequate risk-mitigation strategy. Aptar CSP's challenge as an active material science company is to provide pharma companies with an active packaging-based solution to prevent

nitrosamine formation and/or remove nitrosamine impurities post-formation at the point of packaging, reducing the need for changes to formulation or manufacturing processes. The technology could be used as a primary solution or as an additional measure of protection to complement other strategies used to prevent N-nitrosamine formation.

INNOVATIONS IN ACTIVE MATERIAL SCIENCE TECHNOLOGIES

Nitrites are known to form volatile, gaseous $\mathrm{NO_x}$ components, especially through protonation under mildly acidic conditions. In the absence of any additional reactions, such gases will come to an equilibrium in the environment. However, by removing the $\mathrm{NO_x}$ from the environment, the equilibrium reaction will drive the formation of additional $\mathrm{NO_x}$ gases that, if removed, will continue until the trace levels of nitrite are consumed.

New innovations in active material science technologies can offer an active packaging-based solution that can react with NO_x gases in the packaging to inhibit nitrosamine headspace formation. Leveraging over 20 years of material science expertise, Aptar CSP Technologies' 3-Phase Activ-PolymerTM platform technology is delivered in a unique formulation comprised of a base majority polymer that provides the structure, a channelling agent and active particles (Figure 1). The channelling agent coats and distributes active particles throughout the polymer matrix and allows targeted molecules to enter through the hydrophilic channels. Those molecules migrate to the active particles, where they are adsorbed and permanently removed from the headspace.

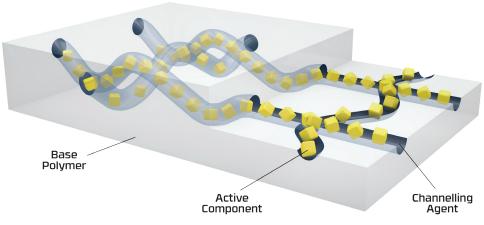


Figure 1: 3-Phase Activ-Polymer™ technology matrix.

For the purposes of this study, the technology, deployed as an Activ-FilmTM material sealed together with MCC tablets, was engineered to quickly and efficiently capture and/or react away the volatile NO_x species, thus preventing the formation of nitrosamines by removing the nitrosating agent (Figure 2). This technology is called N-Sorb (Figure 3).

N-SORB STUDIES

In one study, four versions of N-Sorb mitigation films were exposed to MCC tablets in sealed foil bags during a six-day ageing period. The methodology⁵ used for the experiment was based on a chemical derivative method that reacts with all available nitrites directly into a measurable compound. The compound was detected and quantified using a gas chromatographymass spectroscopy headspace, with a detection ranging from 5 to 100 ppm. The instrumentation specifications for the experiment were:

- Agilent GC 8890, Model Number G3542A
- Agilent MSD 5977, Model Number G7077C
- Agilent Headspace 8697, Model Number G4511A.

All four versions of the N-Sorb mitigation film significantly reduced the average levels of nitrite present in the samples compared with the controls. None of the mitigant's range of concentration overlapped or extended into the control concentration, indicating that nitrite levels were significantly reduced after ageing. After six days of ageing at 60°C in an oven, the samples containing the MCC and either N-Sorb 20 or N-Sorb 23 mitigation film had the lowest level of nitrite concentration remaining in the headspace (Figure 4).

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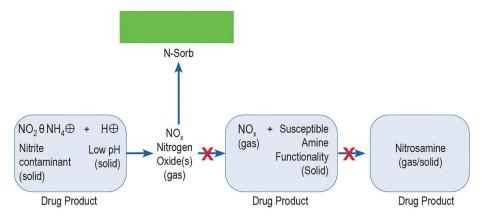


Figure 2: How N-Sorb can prevent nitrosamine formation.4

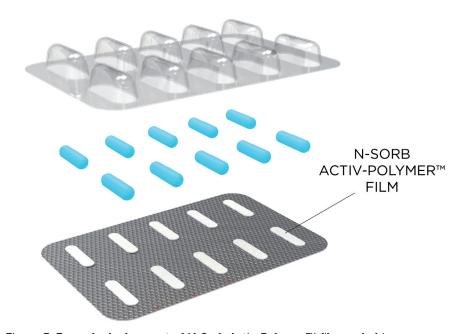


Figure 3: Example deployment of N-Sorb Activ-Polymer $^{\text{TM}}$ film sealed in an Activ-Blister $^{\text{TM}}$ configuration with tablets.

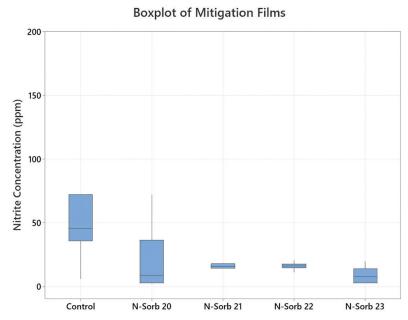


Figure 4: Nitrite concentration (ppm) versus N-Sorb mitigation films sealed with MCC tablets.

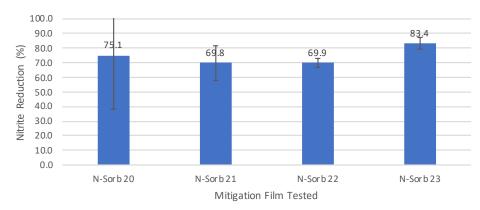


Figure 5: Percentage reduction of nitrites in MCC tablets sealed with N-Sorb mitigation films when compared with control sample.

"Aptar CSP Technologies' innovative Activ-Polymer™ platform with its N-Sorb technology offers pharmaceutical developers a promising alternative to current nitrosamine mitigation strategies."

Nitrite was reduced by all N-Sorb mitigation films tested with MCC tablet samples during the ageing period compared with the amount measured in the control samples (Figure 5). All the mitigation films exhibited a reduction of ~70% or more compared with the amount measured in the control sample. MCC tablet samples exposed to the N-Sorb 23 mitigation film during the test period resulted in the highest reduction of nitrite concentration, along with the lowest variation of the top two mitigation film performers, as illustrated by Figures 4 and 5.

CONCLUSION

Aptar CSP Technologies' innovative Activ-Polymer™ platform with its N-Sorb technology offers pharmaceutical developers a promising alternative to current nitrosamine mitigation strategies by actively removing nitrosamines from the headspace of sealed drug packaging. The key findings from this study demonstrate the potential of N-Sorb as a potent scavenger of nitrite precursors:

- Significant Nitrite Reduction: All tested N-Sorb mitigation films significantly reduced nitrite levels in the headspace compared with controls, with reductions exceeding 70% in some cases.
- Efficient Capture: N-Sorb 23 exhibited the highest nitrite reduction and lowest variation, demonstrating its efficacy in capturing and/or reacting with the

- volatile NO_x species responsible for nitrosamine formation.
- Post-Formation Mitigation Potential:
 While the study focused on preventing
 nitrosamine formation, the technology's
 ability to capture nitrite supports its
 potential efficacy for reducing existing
 NDSRIs in packaged drugs.

For drug formulation and analytical scientists, N-Sorb presents a compelling solution offering several advantages:

- Targeted Intervention: N-Sorb specifically addresses the root cause of nitrosamine formation – the presence of nitrosating agents, such as NO_x gases – without requiring extensive formulation changes.
- Simplicity and Ease of Integration: N-Sorb can be seamlessly integrated into existing packaging formats and processes, minimising disruption to manufacturing workflows.
- Flexibility and Potential for Broad Applicability: The modular design of Activ-Polymer[™] technology allows for customisation to suit specific drug product requirements and target various nitrosating agents beyond NO_v gases.

While further research is warranted to fully validate the long-term effectiveness and compatibility of N-Sorb with diverse drug products, this study provides a strong foundation for its potential as a game changer in the fight against NDSRIs. By offering a

targeted, efficient and easily implementable solution, N-Sorb can empower drug developers to ensure the safety and quality of their products while streamlining compliance with regulatory requirements.

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ABOUT THE COMPANY

Aptar CSP Technologies is part of AptarGroup Inc, a global leader in drug and consumer product dosing, dispensing and protection technologies. Aptar CSP Technologies leverages its active material science expertise to transform ideas into market opportunities, accelerate and de-risk the product development process and provide complete solutions that improve consumers' and patients' lives. The company offers a complete set of services from concept ideation and design and engineering to product development, global production, quality control and regulatory support that results in expedited speed-to-market.

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ABOUT THE AUTHORS



Ivy Comer is Scientist III in the Applications and Research Development Team at Aptar CSP Technologies. She joined the team with a BS in Biochemistry from Auburn University (AL, US), where her passion for science evolved into a deep commitment for innovation, analytical methodologies and meticulous active-applications problem-solving. Ms Comer's aptitude for strategic leadership, communication and collaborative problem-solving has led to impactful solutions within the medical and packaging industries.



Amanda Murph is Scientist II within the Research and Applications Development division at Aptar CSP Technologies. Leveraging her expertise in materials engineering, Ms Murph spearheads the Nitrosamines Mitigation project, where she assumes a leadership role in project planning and execution. Despite her relatively brief tenure in the industry, Ms Murph's commitment to innovation is evident as she continuously explores novel applications for various materials. Driven by a passion for collaborative problem-solving, Ms Murph remains steadfast in her pursuit of excellence, prioritising results and tangible outcomes.

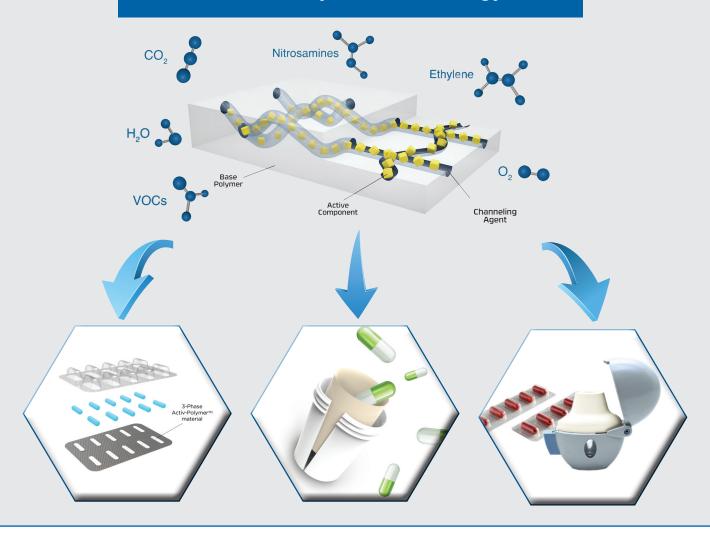


Jason Pratt is Director of Material Science at Aptar CSP Technologies. An experienced scientist with a passion for innovative solutions to commercial opportunities, Mr Pratt is skilled at bringing together and mentoring high power teams to tackle impossible problems. He is a green chemist with over 40 patents across the food, cosmetics, pharma and construction industries.

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